



## Lecture 4

### **Waves in Crystal – continued**

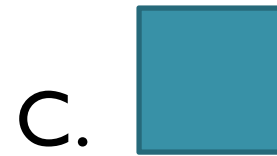
### **Crystal Bonding**

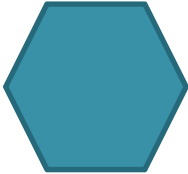
### **Phonons – Soft and Hard**

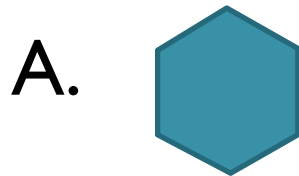
Crystal is nature's diffraction grating  
for any wave, i.e. for any particle!

# Quiz

- Which is most likely the reciprocal unit cell of the unit cell  ?



- Which is most likely the the reciprocal WS cell of the WS unit cell  ?



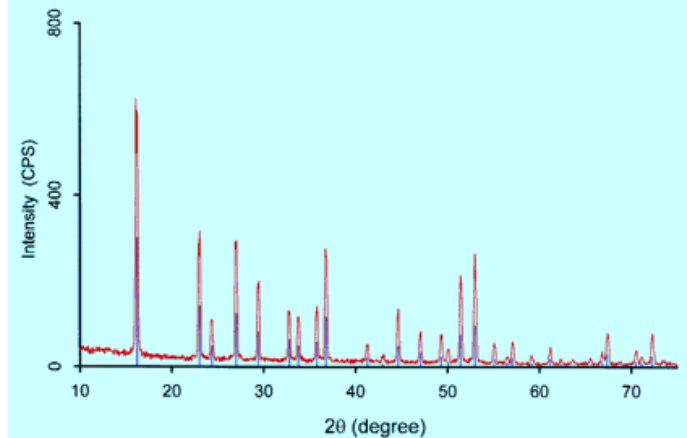
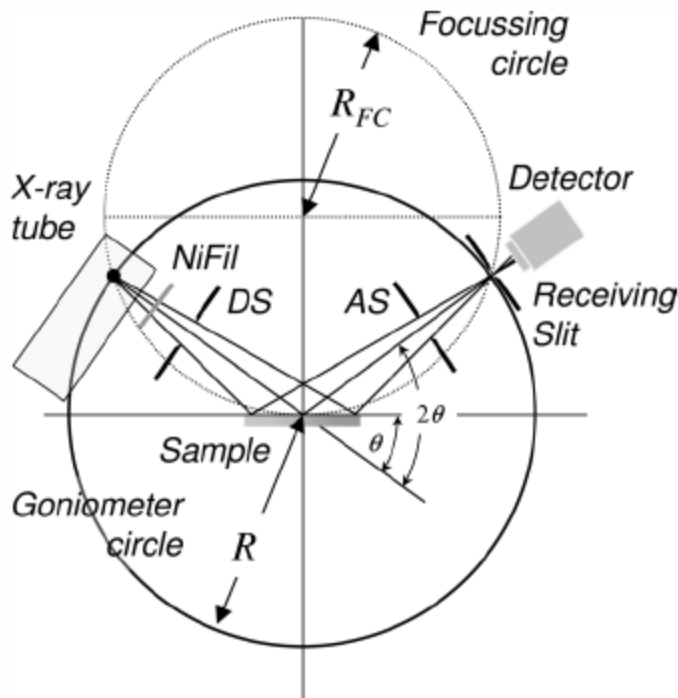
# Quasi-Crystal

- Crystals give Bragg diffraction spots
- But not all materials giving Bragg diffraction spots are crystals
- People suggest that “Quasi-Crystals” can be defined as those non-crystals giving Bragg diffraction spots
- In the language used so far, QC has no Bravais lattice and an infinite size unit cell!
- Nature and human make QCs (penrose tiles, photonic QC)

# X-Ray Diffraction in Practice

## “ $\theta$ - $2\theta$ ” Scan

- Monochromatized or filtered light
- For detailed structure solving
- Can use powder sample

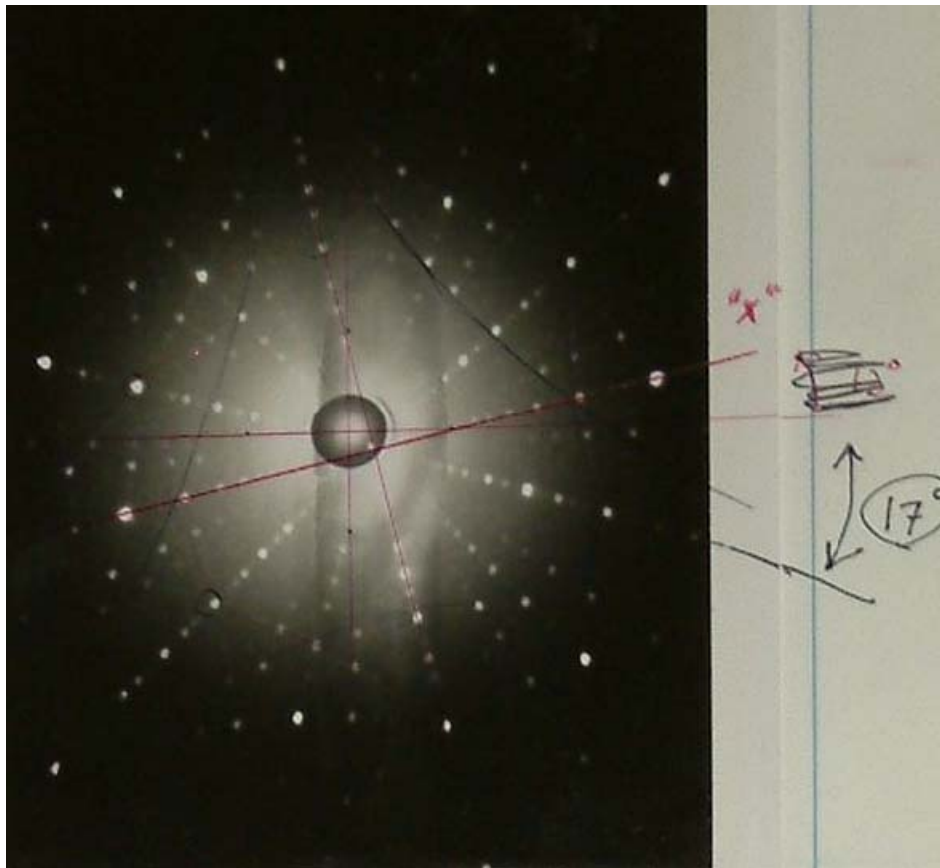


XRD Caption 1: Powder X-ray Diffraction of  $\text{AlF}_3 \cdot 3\text{H}_2\text{O}$  mineral. The vertical lines indicate the positions and peak intensities of the powder diffraction standard from JCPDS database.

<http://www.mastest.com/xrdxrr.htm>

[www.wiley-vch.de/templates/pdf/3527310525\\_c01.pdf](http://www.wiley-vch.de/templates/pdf/3527310525_c01.pdf)

# X-Ray Diffraction in Practice



## Laue method

- Cheap and easy
- Unfiltered white light
- Polaroid film
- Used for single crystal orientation, or check on the single crystallinity

# Structural Factor

- Scattering amplitude (for one atom basis)  
$$= \sum_{\mathbf{R}} \exp(i \mathbf{k}_f \cdot (\mathbf{r} - \mathbf{R})) \exp(i \mathbf{k}_i \cdot \mathbf{R}) f(\mathbf{q})$$
$$\propto f(\mathbf{q}) \sum_{\mathbf{R}} \exp(-i \mathbf{q} \cdot \mathbf{R})$$

- For general case (multiple atom basis)

Scattering amplitude

$$\propto \sum_{\mathbf{R}} \exp(-i \mathbf{q} \cdot \mathbf{R}) \times \sum_m f_m(\mathbf{q}) \exp(-i \mathbf{q} \cdot \mathbf{r}_m)$$

$\mathbf{r}_m$  : position of basis atoms relative to  $\mathbf{R}$

$f_m$  : atomic scattering factor

**S(q)**

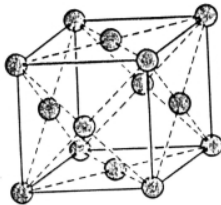
# Crystal Bonding

- Measurement of bonding strength
  - *Cohesive energy*  
Energy to separate crystal into neutral free atoms.
  - *Lattice energy* (ionic crystal)  
Energy to separate crystal into infinitely separated ions (related to “*Madelung potential, energy, constant*”)
- All due to electrons and protons and statistics  
(Many-body electrostatics problem)

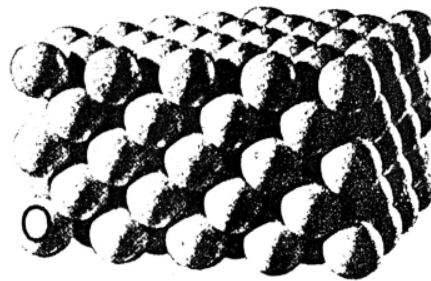
# Crystal Bonding – van der Waals

Inert Gas Elements  
He, Ne, Ar, Kr, Xe

**FCC (CCP)**

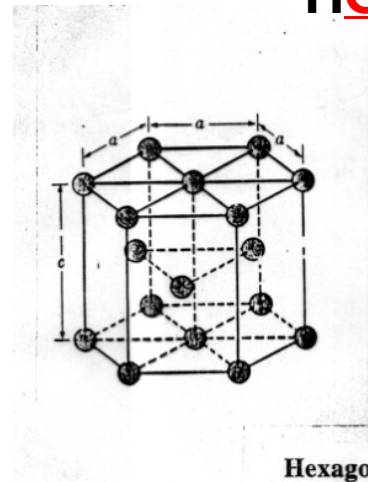


Face Centered Cubic Lattice



Inert Gas Elements  
He

**HCP**



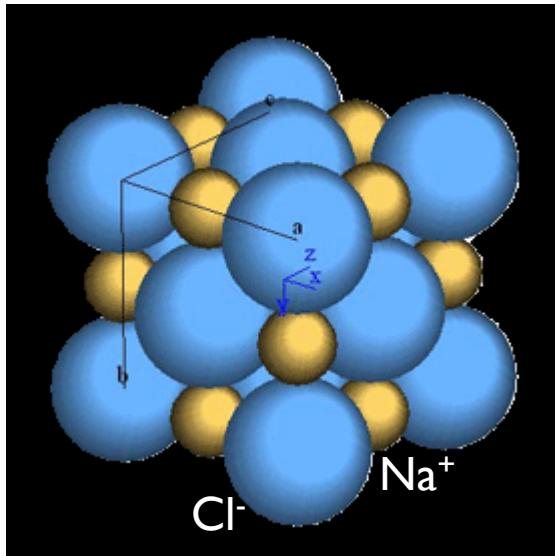
Hexagonal Lattices



<http://www.ae.iitm.ac.in/~sriram/as401/materials>

# Crystal Bonding – Ionic Bonding

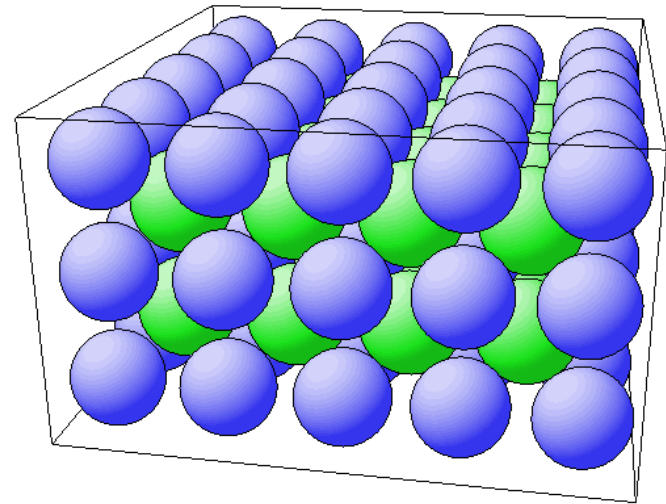
- NaCl



<http://www.matsci.ucdavis.edu/MatSciLT/ENG-45L/images/CaRIne.gif>

Cl<sup>-</sup> bigger than Na<sup>+</sup>, by a factor of 2  
Looks like close packing of bigger ions  
fcc with two atom basis

- CsCl



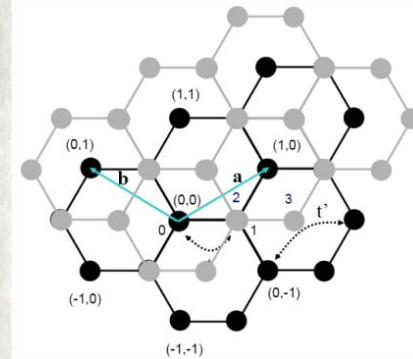
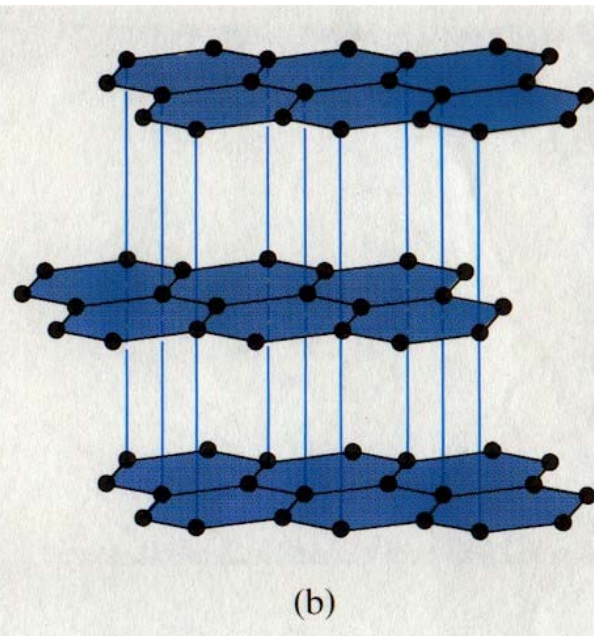
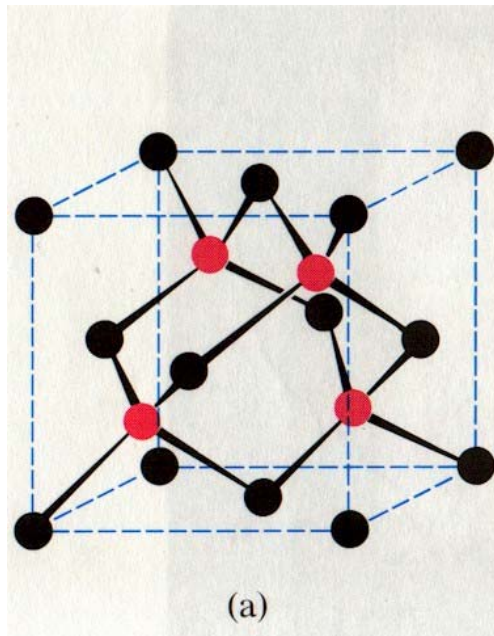
<http://www.cmmp.ucl.ac.uk/~ijf/3c25/CsCl.gif>

sc (simple cubic) with two atom basis

# Crystal Bonding – Covalent Bonding

- Diamond

- Graphite



Top view  
 Gray=top  
 Black=next layer

<http://library.tedankara.k12.tr/chemistry/vol2/allotropy/h76.jpg>

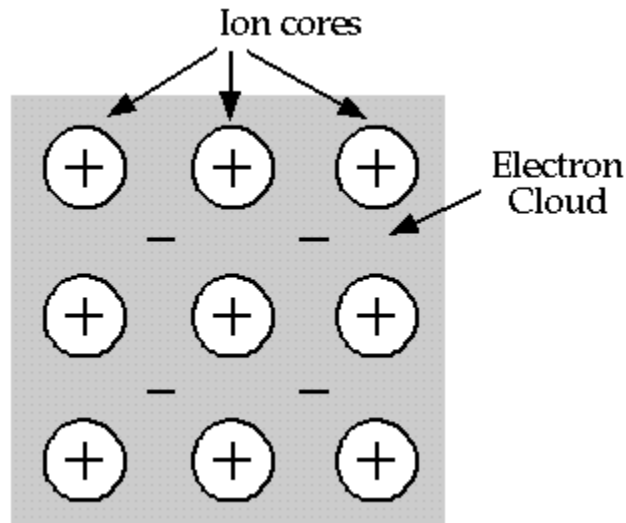
fcc with two atom basis  
 000 and  $\frac{1}{4}, \frac{1}{4}, \frac{1}{4}$   
**ZnS, Si, Ge, GaAs**  
 sp<sup>3</sup> covalent bonding

Hexagonal with four atom basis  
 sp<sup>2</sup> bonding  
 Weak inter-plane bonding (“Van der Waals”)  
 More stable than diamond!

# Crystal Bonding – Metallic Bonding and Hydrogen Bonding

- **Metallic Bonding:** Electrons are released freely from atom and spend most of their time freely. Energy lowering can be viewed as “kinetic energy lowering.”

Na, Cu, Al, ...

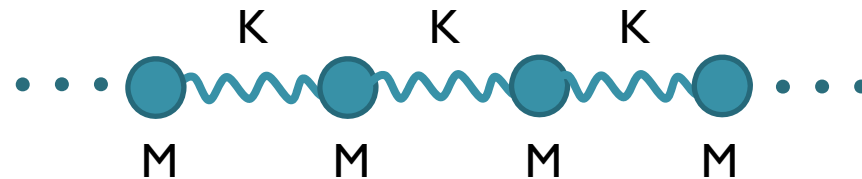


- **Hydrogen Bonding:** DNA, Water (Ice)!

# Crystal Dynamics: Questions

(soft phonon and hard phonon; it is all classical mechanics for now!)

- Consider a monatomic crystal that can be modeled like this with total  $N$  ( $\gg 1$ ) atoms:



1. How many normal modes exist?
2. Consider a long wave length normal mode, in which locally the crystal simply undergoes a uniform translation. What is its frequency as wave length goes to infinity?
3. Consider a mode in which adjacent atoms move exactly in opposite ways. I.e. displacement of  $n$ -th atom is given by  $(-1)^n u$ . What is the frequency?

# Born – Von Karman Boundary Condition



<http://www.mcescher.com/>

# Born – Von Karman Boundary Condition

- Crystal is modeled as Circle (1D), Torus (2D), Hyper-Torus (3D), ...
- Very useful for bulk properties of crystals
- Is not applicable to surface physics or nano-scale physics
- $\sum_{\mathbf{R}} \exp(-i \mathbf{q} \cdot \mathbf{R}) = N \delta_{\mathbf{q}, \mathbf{G}}$  holds rigorously with this boundary condition