



Lecture I

Basic crystallography

Crystals happen

basic emergent phenomenon

“ordering” of many body system

How it happens is hard to know, but one can study what happens – this is the subject of crystallography

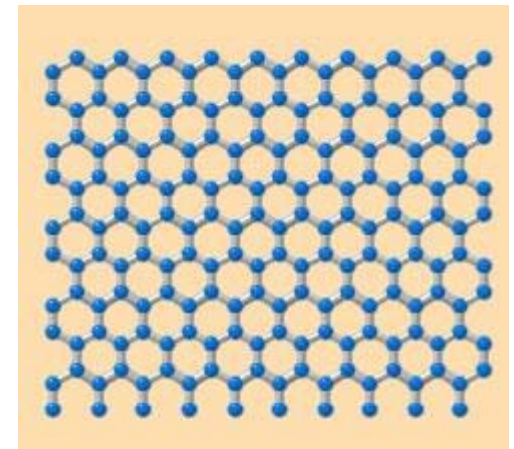
Learning question I

Page 3 of H&H

- Crystal means something that repeats
Something = Basis
Repetition = Lattice

(Bravais Lattice, Crystal Lattice)

- How to define basis and lattice is not unique for a given crystal.
- For the graphene lattice, explain **basis**, **lattice**, and explain whether the lattice is **primitive**.



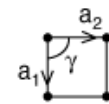
<http://www.ahwahneetech.com>

Learning question 2

Pages 6,7 of H&H

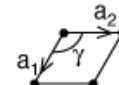
- Explain Figure 1.4 of Hook and Hall.
- Explain what are point symmetries of each lattice. (rotation, reflection, inversion)

- Obtain the **Wigner Seitz unit cell** (see Figure 1.10 and page 13) for each of these, and show that the WS cell has all symmetries noted above.



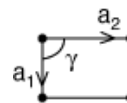
square

$$a_1 = a_2 \quad \gamma = 90^\circ$$



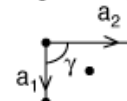
hexagonal

$$a_1 = a_2 \quad \gamma = 120^\circ$$



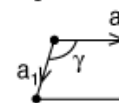
rectangular

$$a_1 \neq a_2 \quad \gamma = 90^\circ$$



centered rectangular

$$a_1 \neq a_2 \quad \gamma = 90^\circ$$

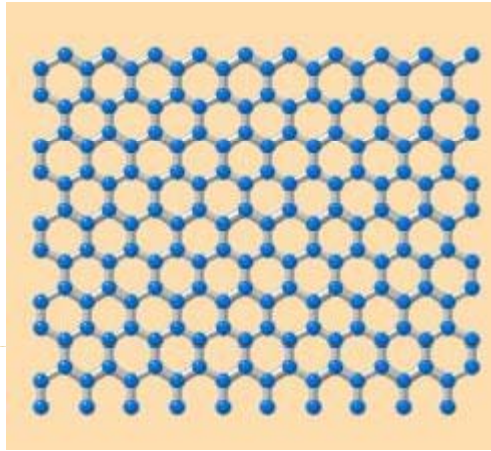


oblique

$$a_1 \neq a_2 \quad \gamma \neq 60^\circ, 90^\circ, 120^\circ$$

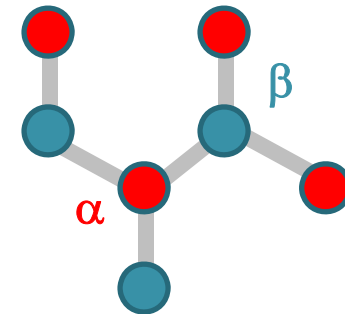
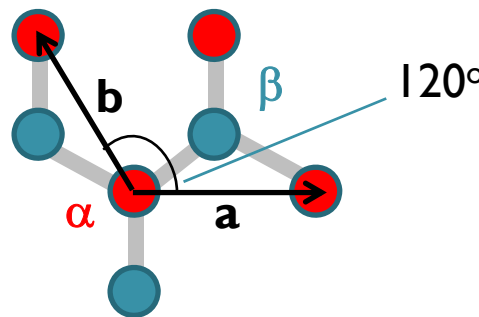
http://whome.phys.au.dk/~philip/q1_05/surflec/fig6_5.gif

Crystal = basis and lattice



“Honeycomb” of graphene
Repeated “Benzene” ring
Each C atom is shared by three rings
Two C atoms per benzene ring
I.e. there are two C atoms per basis
Or, there are two inequivalent C atoms

<http://www.ahwahneetech.com>



Basis = $\alpha + \beta$

Lattice = defined by unit vectors \mathbf{a} , \mathbf{b}

$|\mathbf{a}|=|\mathbf{b}|$, angle = 120° , **hexagonal (or triangular) lattice**

This lattice is a **primitive lattice**, i.e. its basis is not redundant.

Mathematically speaking ...

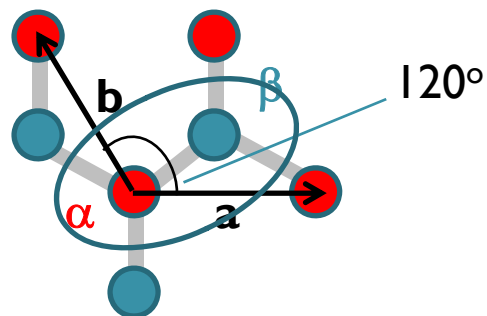
Lattice $\equiv u \mathbf{a} + v \mathbf{b}$, $u, v = \text{integers}$

(For 3d, lattice $\equiv u \mathbf{a} + v \mathbf{b} + w \mathbf{c}$, where $u, v, w = \text{integers}$)

\mathbf{a}, \mathbf{b} = lattice vectors

Basis is then specified by fractional numbers in general

Example:

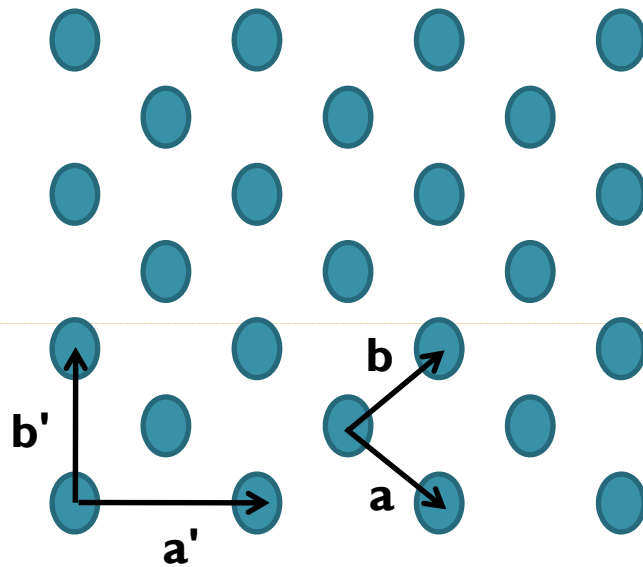


Basis coordinates:

$$0 \mathbf{a} + 0 \mathbf{b}$$

$$(2\mathbf{a} + \mathbf{b})/3$$

Crystal = basis + lattice



Centered rectangular lattice

Primitive lattice is given by,

e.g., \mathbf{a}, \mathbf{b}

$|\mathbf{a}|=|\mathbf{b}|$, angle $\neq 60^\circ, 90^\circ, 120^\circ$

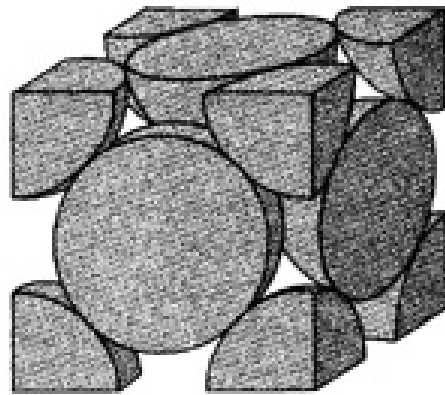
Single atom basis

Or, a lattice can be defined by \mathbf{a}', \mathbf{b}' and two atom basis

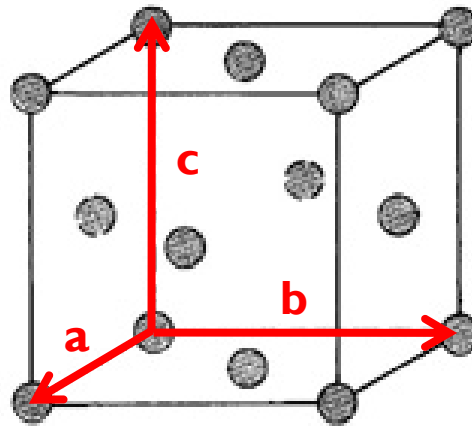
Which lattice to choose is out of convention and preference.
Often a non-primitive lattice used extensively is called a **conventional lattice**.

Conventional lattice and Primitive lattice

Face Centered Cubic Lattice (fcc)



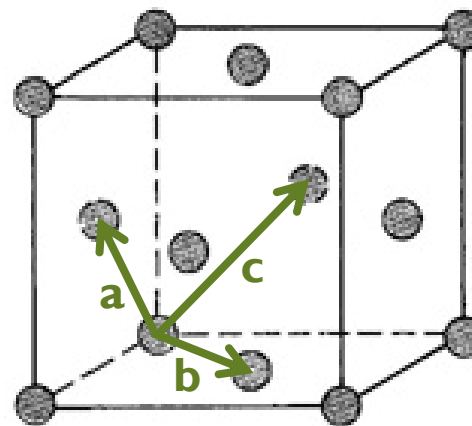
Four Atom Basis



Conventional
(Cube)

$|a| = |b| = |c|$
All angles 90°

Four Atom Basis



Primitive
(Rhombohedron)

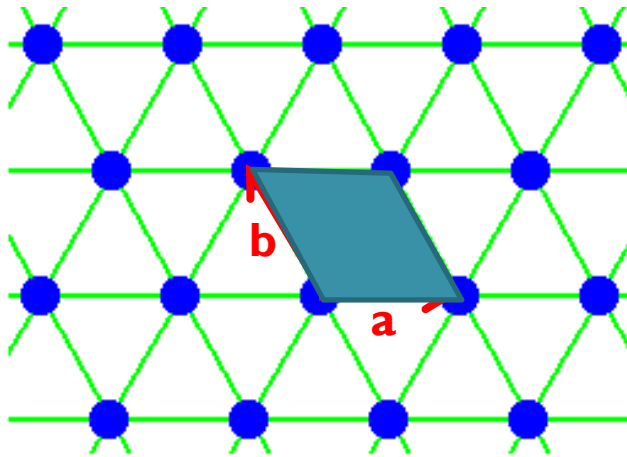
$|a| = |b| = |c|$
All angles 60°

Single Atom Basis

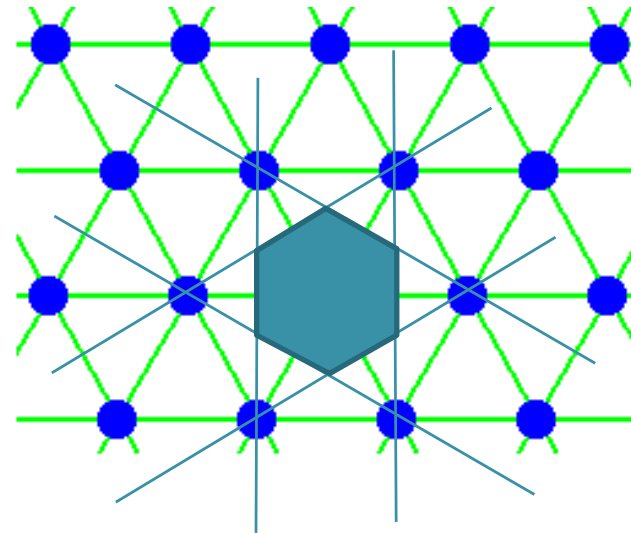
Unit cell

- Repeating geometry assigned to each lattice point

Lattice (NOT crystal) of graphene



A Unit Cell

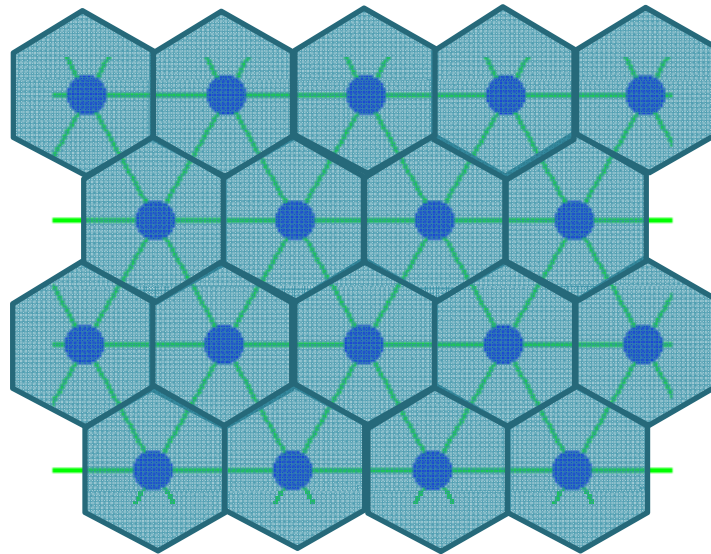


Wigner Seitz Unit Cell
(Hexagon in this case)

http://home.hetnet.nl/~turing/complete_hex_motif_2a.gif

Wigner Seitz cell is the “the most natural” or “the nicest”

- Reflects all symmetries of the lattice itself (why? – too advanced math for this class, but physically one can think of the WS cell as the “imprint” of the environment)



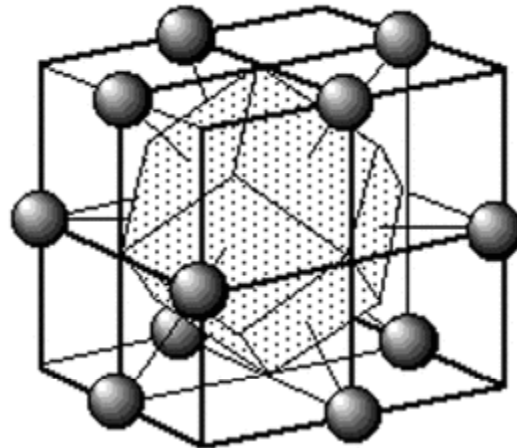
Wigner Seitz Unit Cell

Imagine each blue ball is a balloon (or play dough) which is inflated at the same time. Eventual shape of each ball will be the WS cell – an “imprint” of its neighbors for each lattice point . This is why.

Examples of Wigner Seitz Cells for 3d Lattices

Face centered cubic (fcc)
(note that the origin is shifted)

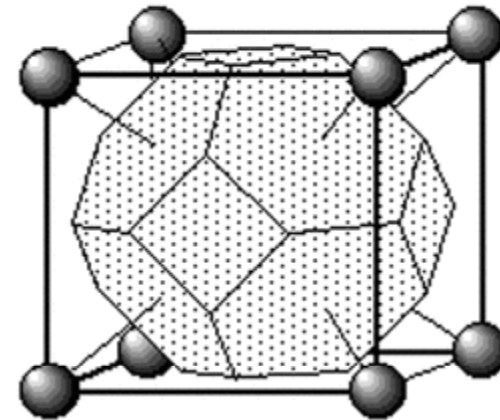
(a)



rhombic dodecahedron

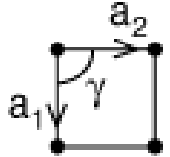

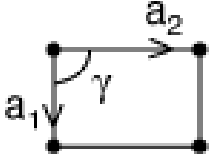

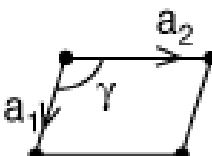
Body centered cubic (bcc)

(b)



truncated octahedron


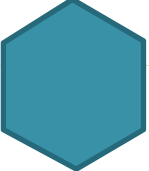



Classification of Lattice (2d)

	WS Cell Shape	Symmetry
	square	4, v, h, i
	hexagonal	6, v, h, i
	rectangular	2, v, h, i
	centered rectangular	2, v, h, i
	oblique	2, i

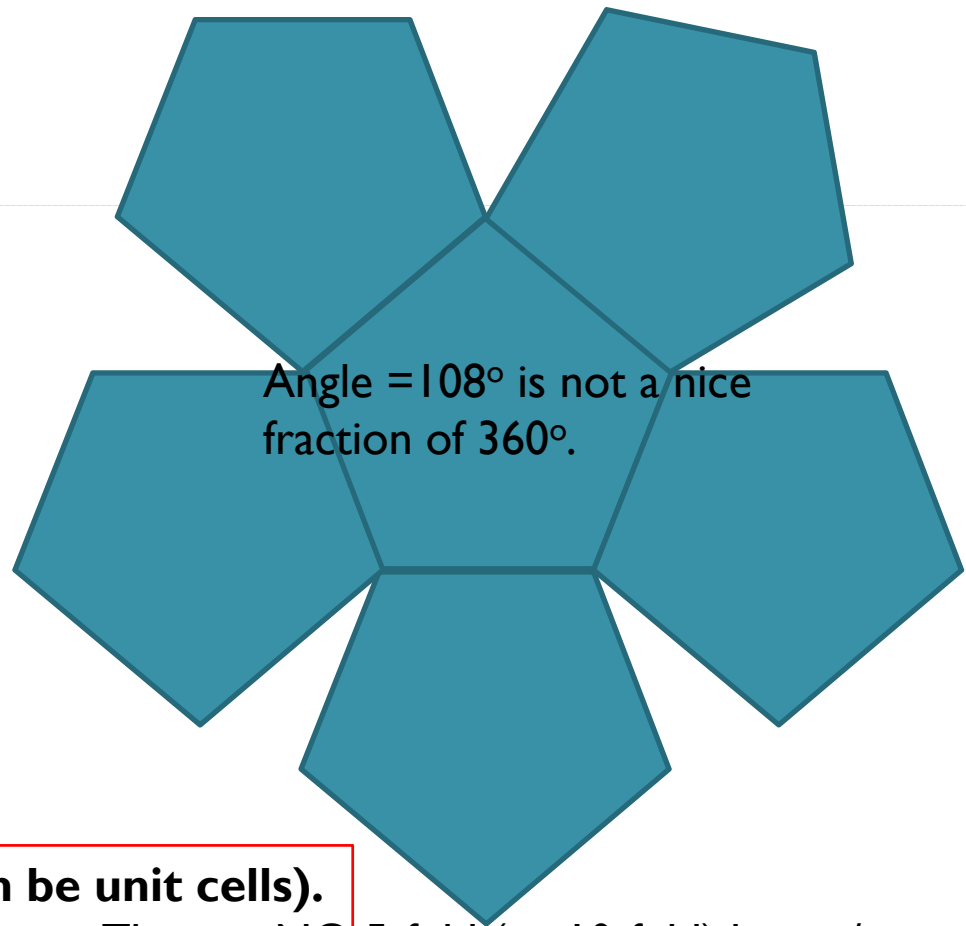
http://whome.phys.au.dk/~philip/q1_05/surflec/fig6_5.gif

2,4,6 = 2,4,6-fold rotation
 v = vertical reflection
 h = horizontal reflection
 i = inversion

These are all WS cell shapes in 2d

WS Cell Shape	Symmetry
 square	4, v, h, i
 hexagon	6, v, h, i
	2, v, h, i
	2, v, h, i
	2, i

Why not (regular) pentagon?

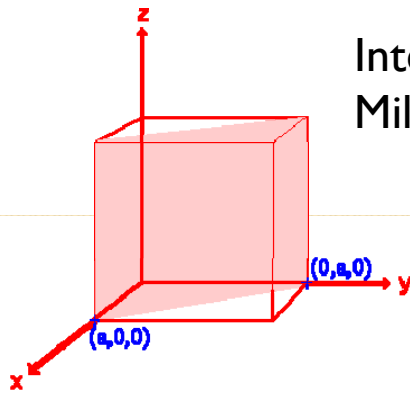


**These can fill space (i.e. can be unit cells).
But a pentagon cannot.**

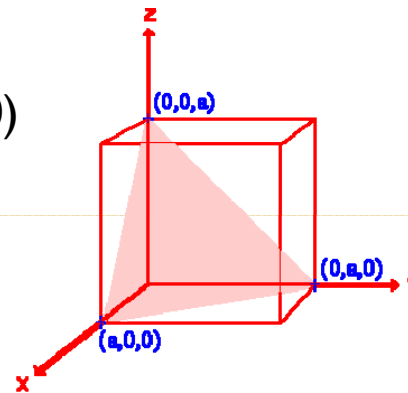
There is NO 5-fold (or 10-fold) lattice/crystal.

Notation for planes (Miller indices) and directions

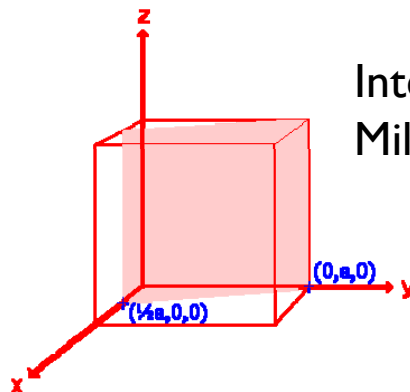
Cubic Lattice Example
(principle the same for all lattices)



Intercepts: $1, 1, \infty$
Miller indices: (110)



Intercepts: $1, 1, 1$
Miller indices: (111)



Intercepts: $1/2, 1, \infty$
Miller indices: (210)

$\{uvw\}$ = group index (all equivalents)

Example:

$$\{1\ 0\ 0\} = (100)(010)(001)(\bar{1}00)(0\bar{1}0)(00\bar{1})$$

bar means minus

Direction notation = $[uvw]$
means $ua + vb + wc$
(u, v, w are integers)

http://www.chem.qmul.ac.uk/surfaces/scc/scat1_1b.htm